ORIGINAL RESEARCH



Enhancing organic waste decomposition with addition of phosphorus and calcium through different sources

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Abstract

Purpose This study assessed the constituent element in rock phosphate (RP) that is responsible for enhancing quick decomposition of cow dung—waste paper mixtures during vermicomposting.

Method Feedstock weighing 5 kg was achieved by mixing 2.16 kg shredded waste paper together with 2.84 kg cow dung sprinkled with water and then enriched with (1% P) as RP from triple superphosphate (TSP), phosphoric acid (PHA) and Ca in the form of CaCl₂ at the level supplied by RP. After mixing, they were loaded into vermireactors and inoculated with matured earthworms at a stocking density of 12.5 g worms/kg feed for the entire 56 days. The decomposition of the mixtures was then monitored by measuring maturity parameters, a germination test for phyto-toxicity and morphological properties was assessed using scanning electron microscopy (SEM).

Result Results revealed that rapid decomposition of waste was recorded where TSP was applied than RP whereas Ca-source had the least effect. A C:N ratio of 12 was achieved within 28 days where TSP was added, while RP, PHA and CaCl₂ needed 42, 56 and more than 56 days, respectively.

Conclusion It was evidently revealed from the results that P was basically the cause of the improved decomposition of the waste mixtures. To achieve quicker and well-humified enriched vermicompost with a C:N ratio of 10 within 20 days of vermicomposting, add water-soluble P sources.

Keywords Vermidegradation · Phosphorus · Enhancement · Organic waste mixtures · Humification

Introduction

Integration of rock phosphate (RP) and P-solubilizing microorganisms into several wastes aimed at nutrient enrichment during vermicomposting have been extensively studied (Mupondi et al. 2010; Biswas and Narayanasamy, 2006; Edwards et al. 2010; Masciandaro et al. 1997; Senesi et al. 1992; Garg and Kaushik 2005; Kaushik and Garg 2004). This is because most waste materials used for composting or vermicomposting (e.g., agricultural wastes, dairy wastes, paper waste, industrial wastes and household wastes), in their natural form, have little nutrient contents of phosphorus and nitrogen among the vital key macro-nutrients required for plant maximum development (Mupondi et al.

In the light of this, Mupondi (2010) during a vermicomposting study mixed RP with shredded waste paper and dairy manure mixtures at proportion greater than 2% P and stated a successfully better decomposition and humification of waste mixture as well as enhanced P availability through increase in the labile pools and less labile pools of P. This was attributed to a collective effect of decomposition of organic matter together with the dissolution of the applied RP. Mupondi (2010) further detected that, added RP improved the overall N content of the vermicomposting mixture with time and attributed it to the absorption influence produced by weight decrease of the organic material resulting from organic matter decomposition and loss of C as carbon dioxide, joined with improved nitrogen mineralization in the substrate as



^{2010,} Biswas and Narayanasamy 2006, Edwards et al. 2010). Besides, RP according to Premono et al. (1996); Kumar and Narula (1999) cannot be applied directly into the soil because of its low dissolution rate, reactivity and high soil pH

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reported by (Atiyeh et al. 2000) when they included > 4% RP into kitchen waste during composting.

However, according to Singh and Amberger (1990), when RP is mixed with organic solid wastes, it both increased the P content and humification of the vermicompost. In another study, (Mishra 1992) got an exceedingly humus-like manure when they composted vegetable wastes with 25% Mussoorie RP.

Unuofin et al. (2016) explored the effectiveness of adding RP at rates less than 2% P to cow dung—waste paper mixtures. The result revealed that, an improved decomposition and humification of the mixtures could be realized even at 0.5% P rate of RP application, even though faster decomposition as well as maturation was achieved at 1% P. Although, 1% P application rate of rock phosphate is what is vital to produce mature vermicompost; however, application greater than 1% P was desirable for vermicompost-enriched P fertilizer.

Based on the elemental nutrient composition of RP, calcium and phosphorous are the dominant elements in most rock phosphates. It was, therefore, hypothesized that since Ca and P were the dominant nutrient elements in RP, they could be responsible for the observed improved decomposition of the mixtures. Hence, it was examined by comparing disintegration of (1% P) of RP, triple superphosphate (TSP), and phosphoric acid (PHA) with Ca in the form of CaCl₂-enriched waste mixtures.

Therefore, the main objective of this study was to determine if P or Ca in the rock phosphate cause the observed enhanced decomposition of the waste mixtures during vermicomposting. A secondary objective was to determine the extractability of P in vermicomposts treated with the different P sources.

Materials and methods

Site description, wastes and earthworms utilized

The experiment was done at the University of Fort Hare, Alice (32 46'S and 26 50'E) under a room temperature of 25 °C in a closed shaded yard. Waste paper used was sourced from both the faculties and Xerox, the University printing press. Phalaborwa in Mpumalanga Province in South Africa supplied phosphate rock. The University local wormery supplied *Eisenia fetida* earthworms. Manure was collected from the Dairy farm in Keiskammahoek. Soluble P fertilizer (Triple super phosphate) was obtained from a company that sells fertilizer in East London. Merck, a chemicals company in Pretoria, South Africa supplied phosphoric acid and calcium chloride.



Worm containers gauging $0.50 \text{ m} \times 0.40 \text{ m} \times 0.30 \text{ m}$ (length × width × depth) were used. This offered 0.2 m^2 of exposed top surface. Matured worms were introduced at a mass of 12.5 g worms/kg feed (Unuofin and Mnkeni 2014). A C:N ratio of 30 was achieved by adding 2.16 kg shredded paper with 2.84 kg cow dung to get 5 kg (dry basis) feedstock that was placed in vermicomposting boxes to accommodate earthworms need for the whole 56 days. Experimental treatments had three replications in a completely randomized design. Boxes were kept at 80% moisture content by sprinkling water when necessary throughout the vermicomposting period (Reinecke and Venter 1987).

Experimental treatments

Rock phosphate (RP), Triple superphosphate (TSP), Phosphoric acid (PHA) and calcium chloride (CaCl₂) were the treatments used. The P-source treatments were inoculated into the waste mixtures at 1%, P since this gave the ideal rate of RP application that brought about speediest decomposition of the waste mixtures (Unuofin et al. 2016). Calcium chloride was also applied to the mixtures at the level of Ca supplied by RP when applied at 1% P. A control (with earthworm yet no P sources) was included. RP, TSP and PHA were mixed each at a rate of 1% P but CaCl₂ was mixed at the level of Ca provided by RP to every 5 kg (dry matter basis). All treated mixtures were vermicomposted in worm boxes having 0.2 m² uncovered top surfaces. Mixtures were inoculated independently with various treatments, and mixed with developed worms at a known stocking density.

Physico-chemical analyses

The samples were analyzed for physiochemical characteristics of the organic mixtures using international standard laboratory methods and procedures for total carbon (C), total nitrogen (N), pH, electrical conductivity (EC), volatile solids (VS), and ash content extractable phosphorus P, and humic substances. pH and EC were determined in a 1:2.5 OM:water mixture). The colloidal mixtures were left to stand for an hour after continuous shaking by a mechanical shaker at 230 rpm for 30 min preceding pH and EC measurements. Whilst, sample weight loss (previously oven-dried at 105 °C) upon ashing at 550 °C for 4 h in a muffle furnace was used to determine volatile solids according to (Ndegwa et al. 2000). Truspec CN Carbon/Nitrogen analyser (LECO Corporation 2003) was used to measure Total N and C of the samples. Bray P extractant was used to assess the extractable P of organic wastes. A continuous flow analyzer (San 2++ Skalar CFA, Skalar Analytical B.V. The Netherlands)





was used to estimate the orthophosphate in the extracts through the ammonium molybdate-antimony potassium tartrate-ascorbic acid method.

Humic substances determination

Del Carmen et al. (2006) method for humic acid extraction procedure was used. Firstly, 0.1 M NaOH (1:20 w/v ratio) was used to treat the samples, and thereafter shaken 4 h on a flat agitator. Resultant colloidal mixture was swirled at 8000 rpm or 8000 x g for 15 min. After that, the clear solution was divided into two halves. Walkley and Black (1934) fast titration method was used to analyze the first half for total extractable carbon fraction (C_{EX}) as defined by (Anderson and Ingram 1996); whereas, the remaining half was attuned to a pH 2 by adding concentrated H₂SO₄ and then left to coagulate for 24 h at 4 °C. The resulting precipitates constituted the humic acid-like carbon (C_{HA}) while portion that remained in solution constituted the fulvic acid-like carbon ($C_{\rm FA}$). The CHA was calculated by subtracting the $C_{\rm FA}$ from the $C_{\rm EX}$. The humification ratio (HR) was calculated as HR = $(C_{EX}/C) \times 100$ and humification index (HI) was calculated as HI = $(C_{HA}/C) \times 100$. PI was calculated as the ratio of C_{HA} to C_{FA} (Mupondi 2010).

Phytotoxicity study

Phytotoxicity was assessed through a seed germination test. Aqueous extracts were prepared from the different vermicomposts with distilled water (1:10 w/v) (Unuofin and Mnkeni 2014). The seed germination bioassay for tomato (*Lycopersicon esculentum*), radish (*Raphanus sativus*) and carrot (*Daucus carota*) was assessed by Tim and Tiquia (Tam and Tiquia 1998) in which two portions of Whatman[®] filter paper were set inside a cleaned petri dish and wetted with the vermicompost extracts. Ten seeds of each yield species were set on the filter paper and keep warm for 5 days in the dark. A control was included for each yield species in which the filter papers were wetted with distilled water. Germination index (GI), relative seed germination (RSG) and relative root lengthening (RRE) were determined as follows:

$$= \frac{\text{Number of seeds germinated in the sample extract} \times 100}{\text{Number of seeds germinated in the control}}$$

RRE (%) =
$$\frac{\text{Mean root elongation in the sample extract} \times 100}{\text{Mean root elongation in the control}}$$

GI (%) =
$$\frac{\text{(\% germination)} \times \text{(\% root elongation)}}{100}$$

Table 1 Selected chemical properties of wastes used in the study

Chemical property	Raw material				
	Cow dung	Waste paper			
pН	7.8 ± 0.001	8.2 ± 0.001			
EC (mS/m)	440 ± 0.002	0.18 ± 0.001			
Total N (g/kg)	23 ± 0.002	3 ± 0.002			
Total C (g/kg)	321 ± 0.002	370 ± 0.001			
C:N	13.2 ± 0.002	205 ± 0.001			
Total P (g/kg)	2.8 ± 0.001	0.5 ± 0.001			
Ash (g/kg)	379 ± 0.001	178 ± 0.001			
C:P	110 ± 0.001	740 ± 0.001			

Determination of the morphology of the vermicompost

Scanning electron microscopy (SEM) model JOEL (JSM-6390LV, Japan) was used to take samples imagery. Concisely, the trials were desiccated and pulverised to pass through a 2 mm mesh. A small representative percentage of the trials was smeared with gold and mounted on SEM. Trials were then imaged by scanning them with a high-energy beam of electrons in a raster scan pattern.

Statistical analysis

Result shown here are the means of three replicate (n=3) which were subjected to repeated measure examination of change (ANOVAR). Fisher's protected least significant difference (LSD) test at $p \le 0.05$ was utilized to decide mean separation. The results were analyzed using JMP[®] version 10.0.

Results and discussion

Initial physical and chemical properties of waste paper and cow dung

Table 1 shows selected characteristics of mixtures used. Carbon content was low and nitrogen content was high in cow manure than in the waste paper. The same trend was observed for total P; as a result, cow dung had narrower C:N and C:P ratios than waste paper. The C:N ratio of cow dung and paper waste is found to be very narrow and very wide, respectively, because such ratio is necessary so that resident or inoculated microflora can multiply by producing extracellular hydrolytic enzyme. Most of the time, final C:N ratio of compost depends upon initial C:N ratio of the substrate. Interestingly, C:N ratio of bacterial cell is also approximately 10:1. Cow manure had high EC and ash content compared to waste paper. The higher EC values in cow





Table 2 Repeated measures ANOVA for Ash content, VS, C:N ratio, pH, EC, extractable Bray P, NH₄-N and NO₃-N and PI, HI, HR

	Effect					
	Treatments		Time		Treatments×time	
	$\overline{F}_{(4,50)}$	d	$\overline{F}_{(4,50)}$	<i>d</i>	$\overline{F_{(16,50)}}$	р
C:N ratio (%)	476.431	< 0.0001	2107.199	< 0.0001	89.579	< 0.0001
	507.648	< 0.0001	1881.033	< 0.0001	399.833	< 0.0001
HI (%)	700.819	< 0.0001	2578.356	< 0.0001	673.436	< 0.0001
HR (%)	1064.597	< 0.0001	3399.654	< 0.0001	560.465	< 0.0001
	1.825	< 0.201	1.081	<0.379	0.993	< 0.483
EC (dS/m)	0.1359	< 0.0001	0.007195	< 0.0001	0.002582	< 0.008
Bray 1 P (mg/kg)	8763.3	< 0.0001	17986.31	< 0.0001	2285.007	< 0.0001
Volatile solids (%)	24.954	< 0.0001	1105.3	< 0.0001	5.603	< 0.0001
Ash content (%)	41.623	< 0.0001	853.453	< 0.0001	8.336	< 0.0001
NH ₄ ⁺ -N (mg/kg)	1105.975	< 0.0001	12530.3	< 0.0001	397.681	< 0.0001
NO ₃ "-N (mg/kg)	1297.168	< 0.0001	2142.724	< 0.0001	364.07	< 0.0001

F values and probabilities are shown for each effect. p < 0.05 are significant n = 3, C:N carbon to nitrogen ratio, PI polymerization index, HI humification index, HR humification ratio, EC elec-

dung could be due to the release of salts from the manure with the passage of time. The pH of both material used for the experiment was alkaline; however, higher pH was recorded in waste paper.

Treatment effects on selected compost stabilization and maturity parameters

Table 2 shows the summary of the interaction amongst treatments and time during decomposition of the mixtures. Results indicated that there was pronounced substantial interaction between treatments and time.

Effect of P and Ca bearing sources on the percentage Ash and VS

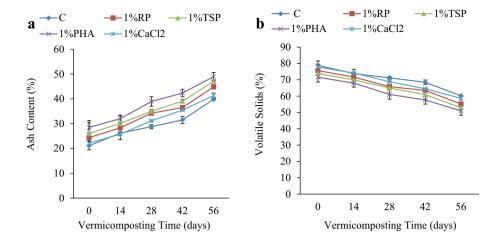
Incorporating P and Ca sources into the mixtures significantly decreased the waste mixture volatile solids and cause a corresponding increase in percent ash of the resultant product; however, this effect was time dependent as shown by a significant treatments × time interaction (Table 2). In all treatments, percent ash increased with time, even though, there were variations in the magnitude among the different treatments (Fig. 1a). Absolute control yielded the lowest values, while the highest values were observed where phosphoric acid (PHA) was added. Generally, the water-soluble P sources resulted in higher percent ash values than the RP, CaCl₂ and control treatments. Gupta et al. (2007); Khwairakpam and Bhargava (2009) reported that during vermicomposting, the concentration of ash increases which proposes that vermicomposting hastens the degree of mineralization and also cuts the amount of water-soluble chemical, which causes likely environmental contamination.

Volatile solids which varied between 71 and 78% in the mixtures, declined amongst initial and latter sampling dates all over the treatments (Fig. 1b). At days 14 and 28, except for PHA-treated waste, reduction in VS values of CaCl2-treated mixtures was not markedly different from the control; whereas, TSP- and RP-treated vermicomposts follow same pattern with a sudden drop between days 42 and 56, respectively. The highest reduction of 50% in VS at day 56 corresponded to PHA-treated product, while the smallest percent ash increase of (35%) occurred in control. Reduction in volatile solids content during vermicomposting of the waste mixtures could be as a result of the disintegration process by microbial activity and loss of carbon in the form of CO₂ (Khwairakpam and Bhargava 2009; Singh and Kalamdhad 2013b). Utilization of carbon by microorganism as a primary energy source for their growth during the vermicomposting process might also cause the reduction of volatile solids (Khwairakpam and Bhargava 2009; Levanon and Pluda 2002). Singh and Kalamdhad (2013a) revealed that OM decreased with the amount of cattle manure inclusion





Fig. 1 Effect of P and Ca bearing sources on the percentage ash and VS



when water hyacinth was subjected to composting. Nevertheless, decrease of organic matter in this study was not subjected to the quantity of waste mixtures but may be because of the presence of easily available food for earthworms.

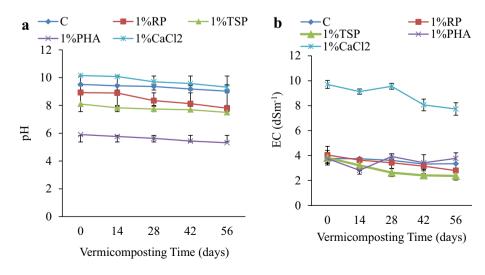
Effect of P and Ca sources on pH and EC

At the beginning of the experiment, vermicomposts pH values ranged between 5.3 and 10.2 (Fig. 2a). The lowest pH was recorded where PHA was applied and the highest where CaCl₂ was added. However, this effect varied with time as shown by the significant interaction between applied treatments and time (Table 2). While most of the treatments showed a significant pH decrease (RP, PHA, and CaCl₂), others (control and TSP) showed no differences over time. The trends in which pH increased during vermicomposting corroborated the work of (Tripathi and Bhardwaj 2004 and Loh et al. 2005) who attributed it to higher fossilization of carbon-based substance as well as (Fares et al. 2005), who associated it with advanced ingestion of organic acids and rise in inorganic constituents of waste by earthworms

activity. However, this present study contradicted the works of (Haimi and Huhta 1987; Ndegwa et al. 2000; Suthar and Singh 2008) in which they observed lower pH values and attributed it to the mineralization of nitrogen and phosphorus into nitrite, nitrates and orthophosphorus as well as the production of CO_2 and organic acids through microbial decomposition.

Electrical Conductivity which varied from 4.1 highest to 3.8 dS/m, increased and decreased over all the treatments between initial and last sampling date from 3.5 to 2.4 dS/m with the highest values of EC recorded where CaCl₂ was applied (Fig. 2b). Two treatments, RP and control shadowed self-same patterns, with abrupt drop between days 42 and 56 (Fig. 2b). However, PHA-treated vermicompost followed a different trend, with lowest EC of 2.8 dS/m at day 14 but gradually increased beyond others except CaCl₂-treated vermicompost at day 28 and thereafter followed similar pattern with RP and control, while TSP-treated vermicompost had the lowest EC values of 2.6 dS/m after day 14 and remained steadily low up till day 56 (Fig. 2b).

Fig. 2 Effect of P and Ca sources on pH and EC





Electrical conductivity as reported by different scientists either increased or decreased during vermicomposting process, some workers reported decrease in electrical conductivity (Garg et al. 2006; Singh et al. 2009) and while others an increase in electrical conductivity (Hait and Tare 2011). The decrease has been attributed to a decrease in ions after forming a complex whereas the increase has been attributed to the degradation of organic matter to release cations and release of different mineral salts in available forms such as phosphate, ammonium, and potassium.

Effect of P and Ca source on the C:N ratio

Added P source and Ca source had significant effects in reducing the C:N ratio of vermicompost, as the effect differed significantly amongst treatments×time (Table 1). The C:N ratio declined meaningfully with time, respectively, at supplementary P-nutrient source; however, there were

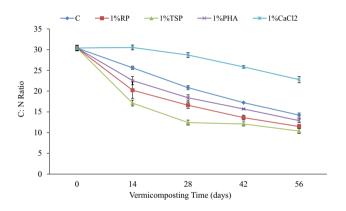
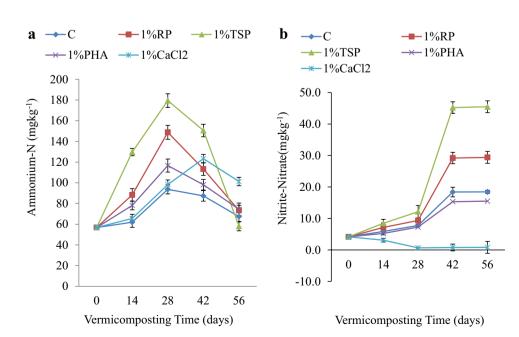


Fig. 3 Effect of P and Ca source on the C:N ratio

Fig. 4 Effects of P and Ca sources on inorganic nitrogen dynamics



obvious differences between different P-source treatments up to 28 days, beyond which differences were minimal except for CaCl2-treated waste mixtures that spread wider apart (Fig. 3). The highest decline in C:N ratio at 14th day corresponded to TSP>RP>PHA>Control, respectively. More decrease in C:N ratio was witnessed past day 28 till the end of the trial but the outcomes of enhanced P-nutrient sources were not meaningfully altered apart from CaCl2-treated vermicompost. Final C:N ratios were in the range of 10-12 in vermicompost treated with P-nutrient source fertilizer, whereas control and CaCl₂-treated vermicompost had C:N ratios of 14 and 22, respectively (Fig. 3). Thus, the addition of P-nutrient bearing source aided reduction of vermicompost C:N ratio and in turn enhanced cow dung-waste paper mixtures vermidegradation at the early stages of vermicomposting up to day 28, beyond which treatment effect was minimal. However, vermicompost treated with CaCl₂ declined minimally but not lower than 22 at day 56 (Fig. 3). Bernal et al. (2009) stated that the drop in C:N ratio might be credited to the breakdown of carbon-based substance because of bacteriological action as well as ammonification and nitrification; both bacteriologically facilitated methods are most powerful all through this period.

Effects of P and Ca sources on inorganic nitrogen dynamics

Table 1 reveals significant treatment × time interaction on ammonium N. A linear increase in ammonium was observed from day zero to day 28 and sharply declined thereafter till termination of the experiment at day 56 (Fig. 4a). The highest increase of about 179 mg/kg ammonium—N was observed in waste mixtures mixed with TSP, whilst the





control samples had the least value of NH_4^+ –N (93 mg/kg). Other treatments increased and followed the order of RP>Control>PHA>CaCl₂.

At day 56, CaCl₂-treated vermicompost had the best NH₄⁺-N content, recommending that vermicomposting procedure was lowest and that Ca had no basic part in the decomposition of the waste mixtures. High NH₄⁺-N concentration at the beginning could be due to N-mineralization, showing the dynamic of OM change and the incidence of unstable medium (Bernal et al. 1998; Zucconi and de Bertoldi 1987). However, many researchers have reported that NH₄⁺-N concentration during vermicomposting usually decreased to reach stability level in final analysis (Mupondi et al. 2010; Levanon and Pluda 2002). Bernal et al. (1998) reported that reductions of NH₄⁺-N concentration prompted

increments of NO₃⁻–N through nitrification, when temperatures turned out to be more satisfactory.

Nitrate–nitrite in the vermicompost followed the same linearly increasing pattern up to day 14 with higher increment observed where TSP was included, while the lowest nitrate–N substance were recorded with compost treated with CaCl₂ till 28 days (Fig. 4b). A sharp increment was noticed from day 28 till day 42 in all the treatments except from mixtures blended with CaCl₂. But, at day 56, TSP, RP, Control and PHA treatments increased linearly with TSP having the highest nitrate concentration of (49 mg/kg), RP (30 mg/kg), Control (20 mg/kg) and PHA (13 mg/kg), respectively. Although Bernal et al. 1998 reported a noteworthy decrease in NO₃⁻–N when wood shavings were added to vermicompost and attributed it to increased TOC/TN which favoured microbial immobilization, nevertheless,

Fig. 5 Effect of P and Ca source on the humification parameter

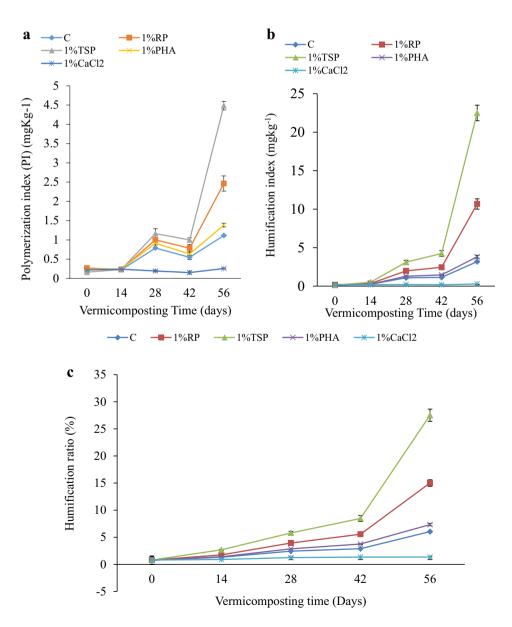




Table 3 Relationship between different P-fertilizer source and Bray P extractable during vermicomposting of cow dung-waste paper mixtures mixed with rock phosphate

P-fertilizer source (%)	Regression equation	R^2	Rate of Bray 1 P release (mgP/kg/ day)		•	Net Bray 1 P increase from day 1 (mg/kg)
Control	$Y = -0.3277X^2 + 25.213X + 75.061$	0.8855	6.7d	172.7	173	93
RP	$Y = -0.1342X^2 + 11.678X + 40.883$	0.5226	11.7c	208.3	208.3	128.3
TSP	$Y = -0.1123X^2 + 10.089X + 42.893$	0.4802	24.7a	424.8	425	345
PHA	$Y = -0.0645X^2 + 6.6401X + 52.538$	0.4644	14.0b	192.2	192.2	112.2
CaCl ₂	$Y = 0.033X^2 - 2.615X + 74.653$	0.8743	2.0e	28.1	28.1	-51.9
CV (%)			5	5	5	
p value			< 0.001	< 0.001	< 0.001	

Numbers followed by different letters in each column are significantly different according to $p \le 0.05$

the increase in N was not immediate and could be as a result of the decline in desiccated mass of OM in the medium because of the breakdown by earthworms as also reported by (Gomez-Brandon et al. 2008). Another reason behind this augmentation could be the closeness and activity of worms in the substrate and discharge of chemicals as reported by (Mupondi et al. 2010).

Effects of P and Ca source on extractable Bray P

Addition of treatments into waste mixtures had significant effect on the amount of P discharge. However, it differed significantly between treatments × time relationship (ANO-VAR) during the study period (Fig. 5). The highest initial extractable Bray P (713 mg/kg) corresponded to the TSP-treated waste mixtures and the lowest to CaCl₂-treated waste mixtures (Table 3). In all the treatments, the release of P by each applied treatments (TSP, RP, PHA and control) either increased or decreased with time and at some points remained linear. At day 14, PHA-treated samples had the highest increment (464 mg/kg) in extractable Bray P compared with TSP-treated waste (328 mg/kg). In case of RP and Control, they followed similar pattern from the beginning till 56th day, while CaCl₂-treated vermicompost remained the lowest up to termination of the experiment (Table 3).

This increase is due to the utilization and mineralization of phosphorus by microbes and enzymes living in the earthworm's gut (Suthar and Singh 2008), as well as OM mineralization in the substrate. Likewise, (Jadia and Fulekar 2008) stated a rise in nitrogen and phosphorus level and a decrease in C/N ratio. But (Parvaresh et al. 2004) observed a decrease in P and ascribed it to inorganic P assimilation freed from the earthworms' tissues, variation of the process, feature of ingredients eaten by earthworms and the conditions of the experiment.

According to (Bhattacharya and Chattopadhyay 2002), phosphate bacteria enable the dissolution of P from P-bearing minerals through the production of phosphatase

enzymes. Discharge of P is facilitated via phosphate chemicals provided through microorganisms living in earthworm's gut and cast (Yan et al. 2012).

Effect of P and Ca source on the humification parameter

Humification parameters [polymerization index (PI), HI, and HR] of waste mixtures were affected by all treatments during vermicomposting period. These effects were similar to the trend observed in C:N ratio as shown in (Table 1). At the beginning of vermicomposting up to 14 days, all treated waste materials had the same PI of 0.2 mg/kg (Fig. 5a). However, as vermicomposting progressed (28 days), changes in PI were noticed. Waste mixtures that received TSP had the highest significant (p < 0.05) increase, whereas waste mixtures to which Cacl₂ was added had the least PI of 0.3. Substantial changes in PI were observed at day 56: waste mixtures mixed with TSP had highest PI value of 4.5 followed by RP (2.5), while control and PHA had equal PI value of 1.2, respectively, but CaCl₂ had 0.1 separately. Similarly, humification indexes (HI) from day zero to day 14 of vermicomposting were not different when compared with the control (Fig. 5b). However, from day 28 to day 56, waste mixed with TSP had the highest HI values of (4.5 and 22 mg/kg, respectively), whilst the least HI was recorded in cow dung-waste paper mixed with CaCl₂ and in the control. The added P-nutrient sources followed the same trend for humification ratio (HR) in vermicompost from the beginning to the end of the study period (Fig. 5c).

The ratio $C_{\rm HA}/C_{\rm FA}$, also known as "degree of polymerization or polymerization index," increases during the composting process, reflecting the formation of complex molecules $(C_{\rm HA})$ from simpler one $(C_{\rm FA})$ and the occurrence of synthesis of organic compounds resistant to microbial degradation (Plaza et al. 2008). Roletto et al. (1985) used the ratio of $C_{\rm HA}/C_{\rm FA}$ to establish the humification level of the OM of compost from different origins, including manure, indicating





Table 4 Effect of treatments on the phytotoxicity of cow dung—waste paper vermicompost

Treatment	Tomato			Carrot			Radish		
	RRE (%)	RSG (%)	GI (%)	RRE (%)	RSG (%)	GI (%)	RRE (%)	RSG (%)	GI (%)
С	80.0c	113.4c	89.5d	90.8ab	108.3c	97.4c	95.6b	91.7c	87.4c
RP	77.1d	133.4b	108.8c	99.3ab	108.3c	120.6b	91.2b	91.7c	95.6b
TSP	125.7a	143.4a	155.6a	109.3a	133.4a	129.9a	102.1a	108.4a	103.2a
PHA	111.6b	106.7d	115.7b	93.4ab	125.0b	115.2b	85.7c	104.2b	91.2c
CaCl ₂	75.8d	90.0e	69.6e	84.8b	91.7d	75.8d	77.7d	75.0d	60.1d
CV (%)	8	6	7	6	5	4	7	7	7
p value	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001

Numbers followed by different letters in each column are significantly different according to the LSD test at $p \le 0.05$. Relative root elongation (RRE), relative seed germination (RSG) and Germination index (GI)

a value greater than 1 for a good humification degree. However, (Senesi 1989 and Iglesias-Jimenez and Perez-Garcia 1992) have suggested that a ratio $C_{\rm HA}/C_{\rm FA}$ greater than 1.6 indicates a good maturity degree achieved by the organic matter. The ratio $(C_{\rm HA}/C_{\rm FA})$ slightly decreases during the vermicomposting time, reaching a value of 1.5, suggesting a roughly reduction in the contents of molecules with a high degree of condensation and humification, which could indicate that the earthworm activities led to a homogeny degradation of all types of C compounds without selectivity, however, maturity level reached do not decrease.

Effect of P and Ca sources on the phytotoxicity

None of the extracts from the final vermicomposted products treated with P or Ca sources had any constrain on seed sprouting of test crops (Table 4). In tomato, each treatment increased the RSG except where PHA and CaCl2 were added. The highest percent of RSG was found with vermicompost mixed with TSP. The same statistically significant (p < 0.0001) trend of RSG was observed for radish and carrot except where CaCl₂ was added. Tomato had the main important value of 125.7% for RRE trailed by radish 109%, whereas the least percent of 102% was observed with carrot upon addition of TSP (Table 4). All tested crops had GI of more than 80% on vermicomposts treated with P but less than 80% where CaCl₂ was applied. This can be because of well advance of decay of natural substrates and diminishment of phytotoxic mixes coming about because of vermicompost maturing. All extracts used for seed germination test were freed of phytotoxicity except were CaCl₂ extract was used because of high EC in regard of Ca-treated product and resilience of tested seeds to salinity (Tiquia 2010). Our results correspond to (Paradelo et al. 2008) who said that GI between 50 and 80% mean moderate phytotoxicity. Likewise, the more than 80% GI watched for all tested seeds in this examination showed that addition of P as TSP, PHA and RP with E. fetida produced fertilizer product freed of phytotoxins as reported by Zucconi et al. (1981 and Tam and Tiquia (1998). Bustamante et al. (2001) also linked GI

of \geq 80% to the absence of phytotoxins in composts. Of the three P sources tested, water-soluble P vermicomposts had the highest germination indices for all crops tested indicating the superiority of these vermicomposts.

Morphology of vermicompost revealed by scanning electron microscopy

(Figure 6c-f) confirmed the degree of humification of the resultant vermicompost from SEM results. At the beginning of the experiment, the waste mixture had a compressed group both of roughage and protein fibres morphologically (Fig. 6a) which were disintegrated by the activities of microbes in enzyme's gut after 56 days (Fig. 6b). However, where earthworms and TSP, RP, PHA were added, highly degraded, fine grain texture vermicompost was produced from the grinding nature exhibited by the earthworm (Fig. 6a-f) except for CaCl₂-treated vermicompost (Fig. 6f). The level of grinding activities of the waste materials by earthworms became more intensified with treatments. Vermicompost produced from TSP (readily available P) enrichment had well-humified and high aggregate particles which was evident in the humification parameters, showing the importance of P rather than Ca (CaCl₂) during bioconversion of waste. Additional affirmation of intense mixtures decomposition is also revealed from SEM pictures (Fig. 6a, c), which indicated that RP addition escalated the vermidegradation of mixtures, and that addition of 1% P of RP brought about reliably more prominent waste conversion during vermicomposting period with the other RP treatments. The resultant SEM photo of the vermicompost exhibited a specific physical appearance that portrayed a scattered separated minute in nature contrast to the control. Thus, SEM images in this study were similar to that of (Lim et al. 2015, 2016), and it reveals that increase in the decomposition of the mixtures significantly depends on time. The wider contrasts in the degree of decomposition observed from various treatments at 14th and 28th days agreed with the time of most extreme microbial action (Unuofin et al. 2016).



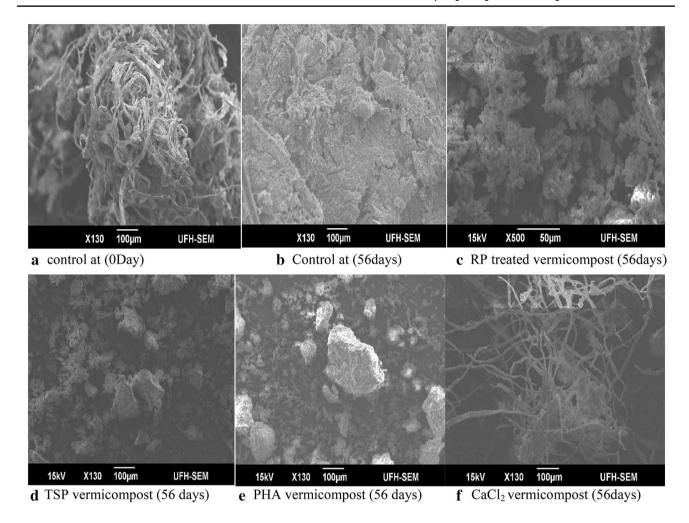


Fig. 6 Scanning electron microscope pictures showing P and Ca source effects on vermicompost morphological properties

Conclusion

The results of this study have demonstrated that phosphorous and not calcium is responsible for the enhanced biodegradation of waste mixtures mixed with P bearing like rock phosphate during vermicomposting. However, the use of watersoluble P sources to enhance vermicomposting may not be justifiable where impure and less expensive P sources such as rock phosphates are available as their use can result in equally mature and P-enriched vermicomposts in 6–8 weeks. The effectiveness of added P in enhancing vermicomposting appears linked to its ability to stimulate microbial growth, C:N ratio decrease, rise in PI, HI and HR. However, it could also be partially linked to meeting the P nutritional requirements of the earthworms. This latter effect will need to be explored in future studies.

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Compliance with ethical standards

Conflict of interest The authors indicated that there is no conflict of interest.

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